

Withanolides from *Withania adpressa*

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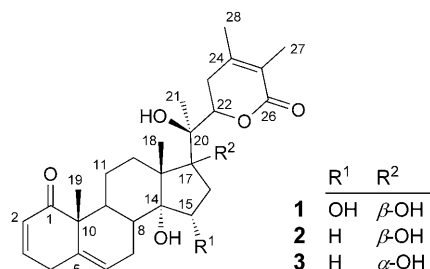
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From the leaves of *Withania adpressa*, a plant endemic to Sahara of Morocco and Algeria, the novel steroidal lactone (22*R*)-14 α ,15 α ,17 β ,20 β -tetrahydroxy-1-oxowitha-2,5,24-trien-26,22-olide (= (15*S*,17*S*)-14,15,17,20-tetrahydroxy-22,26-epoxyergosta-2,5,24-triene-1,26-dione; **1**), was isolated, along with three known compounds, withanolides F (**2**), J (**3**), and oleanolic acid. Their structures were mainly solved by in-depth 1D- and 2D-NMR (including ADEQUATE) experiments, as well as by HR-MS analyses and chemical evidence.

Introduction. – The genus *Withania* (Solanaceae) is known for elaborating withanolides, which are steroidal lactones characterized by a C₂₈ basic skeleton. Several of these substances have displayed various biological activities, such as cytotoxic [1], anticancer [2], immunosuppressive [3], adaptogenic [4], and antifeedant [5] properties. In Morocco, the genus *Withania* is represented by three species: *W. somnifera* (L.) DUNAL, *W. adpressa* COSS., and *W. frutescens* (L.) PAUQUY [6]. *W. adpressa* is a green-grayish shrub, with coriaceous, glaucous leaves; it is endemic to Sahara of Morocco and Algeria [7].

According to the literature, *W. adpressa* has not previously been studied phytochemically. The present investigation describes the isolation and structural elucidation of a new withanolide, identified as (22*R*)-14 α ,15 α ,17 β ,20 β -tetrahydroxy-1-oxowitha-2,5,24-trien-26,22-olide (= (15*S*,17*S*)-22,26-epoxy-14,15,17,20-tetrahydroxyergosta-2,5,24-triene-1,26-dione; **1**). This new compound was isolated together with withanolide F (**2**), withanolide J (**3**), and oleanolic acid from the leaves of the title plant.

Results and Discussion. – Compound **1** was obtained as a colorless solid. Its molecular formula, C₂₈H₃₈O₇, was deduced from the quasi-molecular ion peak at *m/z* 509.2517 ([*M*+Na]⁺) by means of high-resolution electrospray-ionization time-of-flight mass spectrometry (HR-ESI-TOF-MS). Liquid-chromatography ion-trap mass spectrometry (LC/IT-MS) showed the pseudo-molecular ion at *m/z* 504.0



($[M + H_2O]^+$), with a fragment ion at 169.2 ($C_9H_{13}O_3$) due to cleavage of the C(17)–C(20) bond, confirming the presence of one OH group as well as a six-membered lactone substituted at C(20) of the steroidal skeleton. The second moiety of the molecule, after losing an OH group, gave rise to the base peak at m/z 300.9. The most-prominent ions were due to the gradual loss of four molecules of H_2O : m/z 486.0, 468.8, 451.4, 433.1, and 283.3, confirming the presence of four OH groups.

The UV spectrum of **1** showed a strong absorption at λ_{max} 222 nm, characteristic of α,β -unsaturated C=O and α,β -unsaturated lactone chromophores [8]. The IR absorptions at 3416, 1685, and 1660 cm^{-1} suggested the presence of OH groups, a six-membered cyclic ketone, and an α,β -unsaturated lactone, respectively [9].

The 1H -NMR spectrum of **1** (Table) showed five Me *singlets* at $\delta(H)$ 1.15, 1.22 (2 Me), 1.75, and 1.88. The appearance of Me(21) as a *singlet* was a strong indication that the neighboring C(20) was a quaternary C-atom, and the appearance of C(21) at low field ($\delta(C)$ 1.22) confirmed that an OH group was present at C(20). The downfield shift of Me(27) and Me(28) in the 1H -NMR spectrum suggested the presence of a C=C bond between C(24) and C(25), *i.e.*, the presence of an α,β -unsaturated lactone [10]. H–C(22) coupled only with CH_2 (23), and appeared at $\delta(H)$ 4.68 (*dd*, $J(22,23a)=12.7$, $J(22,23b)=3.2$ Hz). H_a –C(23) coupled with H_b –C(23) (16.0 Hz) and with H–C(22) (12.7 Hz), and appeared as a broad *doublet*; and H_b –C(23), coupling with H_a –C(23) (16.0 Hz) and H–C(22) (3.2 Hz), appeared as a *multiplet*.

Three downfield signals at $\delta(H)$ 5.76 (*dd*, $J(2,3)=9.8$, $J(2,4a)=2.0$ Hz), 6.88 (*ddd*, $J(3,2)=9.8$, $J(3,4b)=4.7$, $J(3,4a)=2.4$ Hz), and 5.57 (*d*, $J(6,7a)=5.2$ Hz) represented three vinylic H-atoms, H–C(2), H–C(3), and H–C(6), respectively. The chemical shift of Me(18) ($\delta(H)$ 1.22) was consistent with the presence of OH groups at positions 17β and/or 14β . The quaternary ^{13}C -NMR carbon signals at $\delta(C)$ 82.1 and 88.0, assigned to C(14) and C(17), respectively, confirmed a $14\beta,17\beta$ -diol structure. The ^{13}C -NMR (DEPT) spectrum of **1** indicated the presence of 28 C-atoms, including five Me, six CH_2 , and seven CH groups, as well as ten quaternary C-atoms. The lowfield signals at $\delta(C)$ 203.1 and 165.9 were due to ketone and lactone C=O groups, respectively. The signals at $\delta(C)$ 127.0, 146.4, and 125.1 were attributed to three vinylic C-atoms, C(2), C(3), and C(6), respectively. The peaks at $\delta(C)$ 135.1, 150.7, and 120.2 were assigned to quaternary vinylic C-atoms, C(5), C(24), and C(25), respectively, and the signal at $\delta(C)$ 80.9 was attributed to C(22). The resonance at $\delta(C)$ 78.2 was assigned to an oxygenated quaternary C-atom, C(20). The signals appearing at $\delta(C)$ 20.0, 18.2, 19.4, 20.2, and 12.1 were assigned to Me(18), Me(19), Me(21), Me(27), and Me(28), respectively.

Table. ¹H- and ¹³C-NMR Data of **1–3**. In (D₆)DMSO (**1, 2**) or (D₆)DMSO/CD₃OD 9 : 1 Solution (**3**). δ in ppm, J in Hz; ergostane atom numbering.

	1		2		3	
	δ(H)	δ(C)	δ(H)	δ(C)	δ(H)	δ(C)
C(1)	–	203.1	–	203.5	–	204.1
H–C(2)	5.76 (<i>dd</i> , $J(2,3)=9.8$, $J(2,4a)=2.0$)	127.0	5.74–5.79 (<i>m</i>)	126.9	5.75 (<i>dd</i> , $J(2,3)=9.5$, $J(2,4a)=2.5$)	127.4
H–C(3)	6.88 (<i>ddd</i> , $J(3,2)=9.8$, $J(3,4b)=4.7$, $J(3,4a)=2.4$)	146.4	6.88 (<i>ddd</i> , $J(3,2)=9.8$, $J(3,4b)=4.9$, $J(3,4a)=2.2$)	146.5	6.86 (<i>ddd</i> , $J(3,2)=10.1$, $J(3,4b)=5.0$, $J(3,4a)=2.5$)	146.8
H _a –C(4)	3.28 (<i>br. d</i> , $J(4a,4b)=21.4$)	32.9	2.23–3.31 (<i>m</i>)	32.8	2.22–3.31 (<i>m</i>)	33.3
H _b –C(4)	2.83 (<i>dd</i> , $J(4b,4a)=21.4$, $J(4b,3)=4.8$)	–	2.83 (<i>dd</i> , $J(4b,4a)=21.6$, $J(4b,3)=4.7$)	–	2.82 (<i>dd</i> , $J(4b,4a)=21.8$, $J(4b,3)=5.0$)	–
C(5)	–	135.1	–	135.2	–	135.8
H–C(6)	5.57 (<i>d</i> , $J(6,7a)=5.2$)	125.1	5.56–5.60 (<i>m</i>)	124.7	5.56–5.59 (<i>m</i>)	124.8
H _a –C(7)	1.99–2.04 (<i>m</i>)	25.5	2.00–2.07 (<i>m</i>)	25.1	1.96–2.01 (<i>m</i>)	24.8
H _b –C(7)	1.91–1.99 (<i>m</i>)	–	1.70–1.75 (<i>m</i>)	–	1.67–1.74 (<i>m</i>)	–
H–C(8)	1.98–2.06 (<i>m</i>)	32.9	1.68–1.74 (<i>m</i>)	36.6	2.02–2.10 (<i>m</i>)	35.9
H–C(9)	2.09–2.17 (<i>m</i>)	35.2	2.02–2.15 (<i>m</i>)	35.2	1.71–1.78 (<i>m</i>)	36.2
C(10)	–	50.3	–	50.1	–	51.5
H _a –C(11)	2.06 (<i>qd</i> , $J(11b,9)=J(11b,11a)=$ $J(11b,12a)=13.4$, $J(11b,12b)=4.7$)	22.7	2.00–2.07 (<i>m</i>)	22.8	2.02–2.08 (<i>m</i>)	22.2
H _b –C(11)	1.53 (<i>d</i> , $J(11a,11b)=15.6$)	–	1.47–1.53 (<i>m</i>)	–	1.49–1.54 (<i>m</i>)	–
H _a –C(12)	2.23 (<i>td</i> , $J(12a,12b)=$ $J(12a,11b)=11.8$, $J(12a,11a)=4.0$)	31.7	1.13–1.19 (<i>td</i> , $J(12a,12b)=J(12a,11b)=12.0$, $J(12a,11a)=4.5$)	30.1	2.13–2.19 (<i>m</i>)	27.1
H _b –C(12)	1.10–1.17 (<i>m</i>)	–	1.13–1.21 (<i>m</i>)	–	1.39–1.44 (<i>m</i>)	–
C(13)	–	53.1	–	53.7	–	50.6
14-OH	5.77 (<i>s</i>)	82.1	5.76 (<i>s</i>)	81.4	^a)	85.3
H _a –C(15)	3.65 (<i>d</i> , $J(15,16a)=5.8$)	74.5	1.32–1.40 (<i>m</i>)	31.9	1.64–1.71 (<i>m</i>)	32.9
H _b –C(15) ^b)	4.60 (<i>br. s</i>)	–	1.52–1.60 (<i>m</i>)	–	1.44–1.51 (<i>m</i>)	–
H _a –C(16)	2.72 (<i>dd</i> , $J(16a,16b)=15.5$, $J(16a,15)=6.6$)	46.4	2.32–2.39 (<i>m</i>)	35.7	2.67 (<i>ddd</i> , $J(16a,16b)=14.2$, $J(16a,15a)=11.7$, $J(16a,15b)=1.9$)	33.6
H _b –C(16)	1.56 (<i>d</i> , $J(16b,16a)=15.6$)	–	1.49–1.55 (<i>m</i>)	–	1.74–1.81 (<i>m</i>)	–
17-OH	4.60 (<i>br. s</i>)	88.0	4.64 (<i>s</i>)	87.3	^a)	87.8
Me(18)	1.22 (<i>s</i>)	20.0	1.00 (<i>s</i>)	20.3	1.02 (<i>s</i>)	18.3
Me(19)	1.15 (<i>s</i>)	18.2	1.14 (<i>s</i>)	18.3	1.15 (<i>s</i>)	18.7
20-OH	6.65 (<i>s</i>)	78.2	6.85 (<i>s</i>)	78.1	^a)	77.0
Me(21)	1.22 (<i>s</i>)	19.4	1.20–1.26 (<i>m</i>)	19.1	1.16 (<i>s</i>)	19.7
H–C(22)	4.68 (<i>dd</i> , $J(22,23a)=12.7$, $J(22,23b)=3.2$)	80.9	4.65 (<i>dd</i> , $J(22,23a)=12.7$, $J(22,23b)=3.6$)	81.0	4.53 (<i>dd</i> , $J(22,23a)=12.9$, $J(22,23b)=3.5$)	80.1

Table (cont.)

	1		2		3	
	$\delta(\text{H})$	$\delta(\text{C})$	$\delta(\text{H})$	$\delta(\text{C})$	$\delta(\text{H})$	$\delta(\text{C})$
H _a -C(23)	2.49 (br. <i>d</i> , $J(23\text{a},23\text{b})=16.0$)	34.3	2.35–2.50 (<i>m</i>)	34.3	2.28–2.35 (<i>m</i>)	31.6
H _b -C(23)	2.35–2.42 (<i>m</i>)	–	2.35–2.50 (<i>m</i>)	–	2.41–2.49 (<i>m</i>)	–
C(24)	–	150.7	–	150.7	–	151.1
C(25)	–	120.2	–	120.1	–	120.3
C(26)	–	165.9	–	165.9	–	166.4
Me(27)	1.88 (<i>s</i>)	20.2	1.85 (<i>s</i>)	20.2	1.88 (<i>s</i>)	20.2
Me(28)	1.75 (<i>s</i>)	12.1	1.72 (<i>s</i>)	12.1	1.73 (<i>s</i>)	12.1

^a) OH Group exchanges in this solvent. ^b) 15-OH for **1**.

The ¹H, ¹H-COSY, NOESY, HSQC, and HMBC spectra allowed us to determine the molecular structure of compound **1**. However, because HMBC data are often compatible with structures where carbon pairs are inverted, ADEQUATE experiments were carried out (Fig.) [11], which are based on the low natural abundance of directly-bound pairs of ¹³C, to unequivocally confirm the structure. The close resemblance of the ¹³C-NMR chemical shifts of rings *C* and *D* in **1** with those of coagulin H [12] was a strong indication that, indeed, the two products differed only in ring *B*. The β -configuration of the OH group at C(14) was excluded since the ¹³C-NMR chemical shifts of known compounds [13–15] with β -configuration were significantly different at C(8), C(14), C(16), and C(18). The α -orientation of the 14-OH group was also supported by an NOE observed with H_a-C(12). Similarly, the α -configuration of the 17-OH group could be excluded because the ¹³C-NMR chemical shifts for C(12), C(13), C(18), C(21), and C(22) were not well-correlated with those reported in the literature [16][17]. Finally, NOE due to contacts between CH₂(12), H-C(22), and CH₂(23) would not be possible if the side chain was β -oriented.

All known 15-hydroxylated withanolides are α -configured [12][14][15]. The scalar coupling constant ³*J*(15,16), with a value of 6 Hz, was taken as a clear indication of the

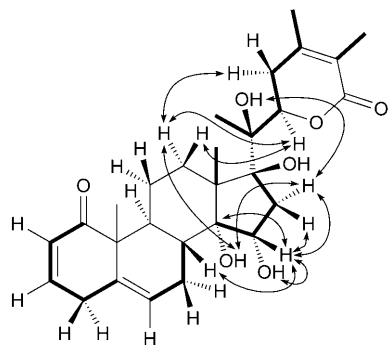


Figure. Key 1,1-ADEQUATE (—) and NOE (↔) correlations of **1**

α -configuration of the 15-OH group [12][13]. Further, an NOE between H–C(8) and H–C(15), both lying above the molecular plane (β), confirmed that the 15-OH function was α -oriented. The configuration at C(22) was determined as (*R*), *i.e.*, the natural configuration in ergostanes, since H–C(22) was found to be a double *doublet* with two coupling constants, which is characteristic for axial–axial and axial–equatorial interactions with CH₂(23) [18].

The IR, UV, ¹H-NMR, and ¹³C-NMR data of compound **2** were very similar to those of **1**, the only difference being the absence of the 15-OH group. The molecular skeleton was easily assembled by means of COSY and HMBC spectra. The configuration of **2** was based on the very close match of ¹³C-NMR chemical shifts with literature data of compounds carrying α - and β -OH groups at C(14) and C(17), respectively [12][16–20]. Compound **2** had been isolated previously from *W. somnifera* growing in Israel, and named withanolide F [19]. However, this is the first report of its isolation from *W. adpressa*.

A set of classical NMR experiments, including ¹H-NMR, DQF-COSY, {¹H}¹³C, DEPT-135, HSQC, and HMBC, allowed us to identify compound **3** as withanolide J [17]. The ¹H,¹H-COSY data and ¹H,¹³C-NMR correlations for **3** were identical with those of withanolide F (**2**). The pattern of differences in ¹³C-NMR chemical shifts between the two withanolides demonstrated that they were 17-epimers [21]. Moreover, both the ¹H- and ¹³C-NMR data of compound **3** successfully passed the test for the determination of the chirality at C(17) developed by Kirson and Gottlieb [16] for a pair of molecules that only differ from ours in a 5 β ,6 β -epoxide instead of a C(5)=C(6) function. Finally, compound **3** was identified as withanolide J by comparison of its ¹³C-NMR data with those published earlier [17].

Oleanolic acid, a pentacyclic triterpenoid sapogenin, was isolated for the first time from the genus *Withania*. It was characterized by comparison of its physical and spectroscopic (¹H-, ¹³C-NMR, DQF-COSY, DEPT-135, HSQC, HMBC) data with those published in the literature [22], and by comparison with an authentic sample.

We thank *André Pinto* for performing most NMR experiments and for spectrometer management. The ADEQUATE experiment was run on the 600-MHz spectrometer located at EPFL, Lausanne, in the frame of the inter-university program 'spectromètre à haut champ d'intérêt romand'.

Experimental Part

General. Thin-layer chromatography (TLC): pre-coated silica gel 60 *F*₂₅₄ plates (*Merck*), visualization under UV light and by spraying with *Dragendorff's* reagent. Column chromatography (CC): regular silica gel (63–200 μ m) or reverse-phase (RP) silica gel (40–63 μ m), *Lobar Lichroprep RP-18* (*Merck*) and *SymetryPrep RP-18* (150 \times 19 mm i.d.; *Waters*). Melting points (m.p.): *Mettler FP800* apparatus; uncorrected. Optical rotations: *Perkin-Elmer 241-MC* polarimeter. IR Spectra: *Perkin-Elmer Spectrum-One-B FT-IR* spectrometer, with KBr pellets; in cm⁻¹. UV Spectra: λ_{\max} (log ϵ) in nm. NMR Spectra: *Bruker DRX-500* apparatus, at 500 (¹H) or 125 MHz (¹³C), or *Bruker DRX-600* apparatus equipped with a low-temperature detection coil optimized for ¹H detection, at 600 (¹H) or 150 MHz (¹³C); δ in ppm rel. to Me₄Si, *J* in Hz. HR-ESI- and APCI-MS: *QStar XL-TOF* and *MAT-LCQ* mass spectrometers, resp., in *m/z*.

Plant Material. The leaves of *Withania adpressa* Coss. were collected in the South of Morocco, April 2003. The plant was authenticated by Dr. Aziz Abbad, Department of Biology, Cadi Ayyad University, Morocco. A voucher specimen (No Mar4223) was deposited at the Herbarium of the Botany Department, Faculty of Sciences-Semlalia, Marrakech, Morocco.

Extraction and Isolation. The dried and powdered leaves (3 kg) of *W. adpressa* were successively extracted with hexane and MeOH (12.5 l each) in a Soxhlet apparatus. After solvent evaporation, the methanolic residue (390 g) was suspended in H₂O (500 ml) and extracted with CH₂Cl₂ (5 × 500 ml). The CH₂Cl₂-soluble fraction (15 g) was subjected to CC (SiO₂; CH₂Cl₂/MeOH 100:0 → 0:100) to afford nine fractions: *Fr. 1–9*. *Fr. 3* (800 mg) was further subjected to CC (1. *LichroPrep*; MeCN/H₂O 10:90 → 100:0; 2. *SymetryPrep*; MeCN/H₂O 4:6) to afford **1** (80 mg). *Fr. 2* (350 mg) was purified by CC (*Lichroprep*; MeCN/H₂O 5:95 → 100:0) to afford eight fractions: *Fr. 2.1* to *2.8*. *Fr. 2.7* afforded **2** (57 mg). *Fr. 1* (1.5 g) was resubjected to CC (*Lichroprep*; MeCN/H₂O 10:90 → 100:0), which gave 62 fractions (*Fr. 1.1* to *1.62*). *Fr. 1.12* yielded withanolide J (180 mg), and *Fr. 1.61* afforded oleanolic acid (210 mg).

(15*S*,17*S*)-14,15,17,20-Tetrahydroxy-22,26-epoxyergosta-2,5,24-triene-1,26-dione (**1**). Colorless solid. M.p. 182–185°. $[\alpha]_{\text{D}}^{25} = +103.3$ ($c=1$, MeOH). UV (MeOH): 222 (4.24). IR (KBr): 3416, 2978, 2942, 1685, 1660, 1450, 1389, 1296, 1143, 1093, 1026, 1001. ¹H- and ¹³C-NMR: see *Table*. HR-ESI-MS: 509.2517 ($[M+Na]^+$, C₂₈H₃₈NaO₇⁺; calc. 509.2516).

Withanolide F (= (17*S*)-14,17,20-Trihydroxy-22,26-epoxyergosta-2,5,24-triene-1,26-dione; **2**). Colorless solid. M.p. 159–160°. $[\alpha]_{\text{D}}^{25} = +205.1$ ($c=1$, MeOH). UV (MeOH): 226 (4.23). IR (KBr): 3424, 2955, 2877, 1686, 1452, 1388, 1322, 1283, 1141, 1092, 1024, 1001. ¹H- and ¹³C-NMR: see *Table*. HR-ESI-MS: 493.2583 ($[M+Na]^+$, C₂₈H₃₈NaO₆⁺; calc. 493.2567).

Withanolide J (= (17*R*)-14,17,20-Trihydroxy-22,26-epoxyergosta-2,5,24-triene-1,26-dione; **3**). Colorless solid. M.p. 180°. $[\alpha]_{\text{D}}^{25} = +316.6$ ($c=1$, MeOH). UV (MeOH): 226 (4.26). IR (KBr): 3426, 2970, 2876, 1683, 1663, 1456, 1410, 1385, 1321, 1139. ¹H- and ¹³C-NMR: see *Table*. HR-ESI-MS: 493.2599 ($[M+Na]^+$, C₂₈H₃₈NaO₆⁺; calc. 493.2567).

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